Prediction of two-dimensional nodal-line semimetals in a carbon nitride covalent network†

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Carbon nitride covalent compounds have emerged as a prominent member of 2D materials beyond graphene. The experimental realizations of 2D graphitic carbon nitride g-C3N4, nitrogenated holey graphene C2N, and polyaniline C3N have shown their promising potential in energy and environmental applications. In this work, we predict a new type of carbon nitride network with a C9N4 stoichiometry from first principles calculations. Unlike common C-N compounds and covalent organic frameworks (COFs), which are typically insulating, surprisingly C9N4 is found to be a 2D nodal-line semimetal. The nodal line in C9N4 forms a closed ring centered at the Γ point, which originates from the pz orbitals of both C and N. The linear crossing occurs right at the Fermi level contributed by two sets of dispersive Kagome and Dirac bands, which is robust due to negligible spin–orbit coupling in C and N. Furthermore, it is revealed that the degeneracy along the high-symmetry path is protected by out-of-plane mirror or C2 rotational symmetry, rather than in-plane mirror symmetry. The chemical potential difference between C and N, as validated by using a single orbital tight-binding model, plays a significant role in forming the nodal ring. Interestingly, a new structure of the nodal line, i.e., a nodal cylinder, is found in momentum space for AA-stacked C9N4. Our results indicate possible functionalization for a novel metal-free C–N covalent network with interesting semimetallic properties.

Introduction

Graphene1 discovered in 2004 has fostered a new research field of two-dimensional (2D) materials, attracting continued research attention. Much effort has been devoted to searching for new 2D materials beyond graphene. Next to C in the periodic table, N appears to be a suitable partner to compose another important set of 2D materials with C. 2D polyaniline with a C3N stoichiometry has been successfully synthesized by a direct pyrolysis of hexaaminobenzene (HAB) trihydrochloride single crystals very recently.2 The C3N monolayer can be considered as N-substituted graphene with uniformly distributed N atoms in an ordered pattern, which possesses an indirect band gap.3 This C–N compound has great potential not only for applications in nano-electronics but also as a functional unit.4–6 In comparison with the hole-free honeycomb C2N monolayer, holey generated C–N covalent networks7–12 have also drawn a great deal of interest lately. Graphitic carbon nitride, g-C3N4, has been extensively studied for decades,13,14 which crystallizes in a porous framework consisting of holes due to large N content.7 Graphitic C2N has a direct band gap and can be potentially utilized in a wide range of energy and environmental applications, including fuel cells, catalysis, and gas production.7–9 Likewise, nitrogenated holey graphene with a C2N stoichiometry has been synthesized by a bottom-up wet-chemical reaction recently.10 This 2D crystalline C2N network, with regular holes, has a large electronic band gap of 1.96 eV, making it a promising candidate material for optoelectronics10 and gas purification.11,12 In addition, plenty of C–N materials with different C/N ratios have been theoretically designed via first principles calculations.15–18

So far, however, there has rarely been any investigation of 2D C–N covalent networks with metallic or better yet nodal-line semimetal (NLSM) features. NLSMs as new states of quantum matter have been intensively studied recently.19–31 Particularly, C-based Mackay–Terrones crystals (MTCs),28 Ir4 classes,32 PbTaSe2,33 anti-perovskite Cu2Pd2N,34,35 and Ca3P2 (ref. 29) have been found to be 3D NLSMs. Attention has also been paid to 2D NLSMs with respect to both experimental research and theoretical design.30–32,14,16 The experimental realization of 2D planar Cu2Si14 was reported to be a NLSM, in which the nodal lines form two concentric loops centered at the Γ point. However, the band touching in Cu2Si is not rightly lying at the Fermi level and...
the existence of two nodal loops makes its experimental characterization more challenging. First-principles predicted that Hg$_2$As$_2$ (ref. 20) and PdS families$^{21}$ were only 2D NLSMs in the absence of spin–orbit coupling (SOC) but became gapped in the presence of SOC.20,21

In this work, we propose a novel covalent network consisting of C and N atoms with a C$_9$N$_4$ stoichiometry from first principles calculations. Most interestingly, it is found to be a 2D NLSM, in contrast to other chemically produced C–N compounds$^{22,23,24}$ as well as COFs$^{30,43}$ which are all insulating. The nodal line in C$_9$N$_4$ forms an ideal single loop centered at the $\Gamma$ point without overlapping with any other bands in the momentum space. Mechanistically, this single loop is revealed to be the crossing of two separate sets of bands arising from two sub-lattices, namely the Kagome and Dirac bands. The interpenetration between Kagome and Dirac (honeycomb) lattices proposed recently can lead to different properties, like 2D topological insulators,44,45 normal insulators,46 half metals,47 Dirac semimetals,48 and 2D NLSMs.49 In these materials, the sub-lattice sites were occupied by single atoms instead of molecular structural motifs, which is different from our case. The negligibly small SOC in C$_9$N$_4$ is significant for the robustness of the nodal line compared with the annihilation in other 2D NLSMs containing heavy metals.20,21,46,47 Moreover, the large chemical potential difference between C and N plays a crucial role in the formation of the nodal line in C$_9$N$_4$, as further confirmed by charge transfer and tight-binding (TB) analysis. Interestingly, there is a new structure of the nodal line formed, i.e. a “nodal cylinder”, in the whole Brillouin zone (BZ) along the $k_z$ direction for AA-stacked C$_9$N$_4$. The gapless linear bands along the high-symmetry path are similar to those in graphene, which indicates higher electron mobility. Moreover, the planar 2D structure with a large surface-to-volume ratio affords plenty of chemically active sites, potentially useful in sensor applications.49 Even in the bulk form with different stacking configurations, the system remains metallic, which is different from g-C$_3$N$_4$.48 Similar to COFs, we propose a new graphene-type network in organic-conjugated polymer chemistry. Our results open an avenue for the design of both 2D NLSM and functional covalent organic frameworks.

**Results and discussion**

As shown in the right panel of Fig. 1, the proposed covalent network has regular periodic nano-holes, in which C and N atoms are represented by black and blue balls, respectively. The hexagonal C rings in the network form a Kagome lattice, which is indicated by dashed green lines connecting each center of C rings, whereas the N constituents form a honeycomb lattice. Overall, C and N atoms share a 2D hexagonal lattice with the same crystal symmetry ($D_{6h}$) as graphene,9 where the optimized lattice parameter is 9.64 Å. In the left panel of Fig. 1, a C ring joining with six N atoms is shown to form this new type of network. This building block is also the key component of the precursor HAB, which is utilized to synthesize crystalline C$_3$N$_2$ and C$_9$N$_4$ compounds. The unit cell of our proposed C–N network, indicated by red lines in Fig. 1, has 18 C atoms and 8 N atoms resulting in a C$_9$N$_4$ stoichiometry. It is worth noting that C$_9$N, C$_2$N, and C$_9$N$_4$ have very similar lattice geometries (Fig. S1 in the ESI†), indicating intimate structural relationships between these monolayer carbon nitrides. The C$_2$N unit cell is $\sqrt{3} \times \sqrt{3}$ times the C$_2$N unit cell by removing a C ring, while the C$_9$N$_4$ unit cell is twice the C$_2$N unit cell by removing another C ring. According to the atomic chemical environment and symmetry, there are two kinds of C and N atoms in C$_9$N$_4$. The optimized C–C and C–N bond lengths are similar to those in C$_2$N,16 which indicates a strong covalency. The perforated covalent network has rich chemically active edge sites with a lone pair, suggesting its potential application in chemical sensors.

The uniformly distributed N atoms in the C$_9$N$_4$ network have an atomic size comparable with that of C and a five-electron valence structure, which makes it a favorable option to realize a strong covalent network. The C$_9$N$_4$ electronic band structure without SOC is shown in Fig. 2(a). There are two touching points denoted by D1 and D2 in the proximity of the Fermi level (set as

**Computational methods**

The structural optimization and electronic band calculations are performed within the framework of density functional theory (DFT) using the Vienna ab initio simulation package.31 The Perdew–Burke–Ernzerhof (PBE) functional of generalized gradient approximation is employed. The projector-augmented wave$^{42}$ method is used to describe the ion–electron interactions. The HSE06 hybrid functional$^{44}$ is used to calculate the band structures to compare with the PBE results. The van der Waals interactions are included in the stacked structures by using a semiempirical DFT-D3 method.25 The plane-wave energy cutoff is set to be 500 eV, and the energy convergence criterion in the self-consistency process is $10^{-6}$ eV. A 20 Å long slab layer is applied along the $Z$ direction to avoid interactions between periodic images. A $\Gamma$ centered $9 \times 9$ grid of $k$ points is used to sample the BZ. The phonon spectrum is calculated using a $2 \times 2$ supercell with the Phonopy$^{46}$ code. Ab initio molecular dynamics (AIMD) simulations with an NVT ensemble are conducted to investigate the thermal stability of the C$_9$N$_4$ monolayer. The molecular properties are calculated using the Gaussian package$^{57}$ at the B3LYP level.

**Fig. 1** The left panel shows the building block for the C$_9$N$_4$ covalent network. The right panel presents the C$_9$N$_4$ crystal structure, where the red line indicates the unit cell. The C rings form a Kagome lattice connected by green dashed lines.
zero), which indicates a semimetallic nature. This unusual feature is apparently different from those of most previously studied C–N compounds\textsuperscript{2,3,7,10} and COFs,\textsuperscript{42} which without exception are semiconducting or insulating. To reveal the origin of these two crossing points, the atomic-orbital resolved band structure is plotted in Fig. 2(a), where the sizes of the circles are proportional to the weight of related orbital contributions. It is clear that the two linear crossing bands near the Fermi level stem from the \( p_z \) orbitals of C and N atoms. The bands corresponding to the \( p_x \) and \( p_y \) orbitals lie much deeper below the Fermi level. Furthermore, there are two sets of distinguishable bands, the Kagome and Dirac bands. The former consists of one \( \delta \) band and two dispersive bands, arising mainly from the \( p_z \) orbitals of C; the latter mainly arises from the \( p_z \) orbitals of both C and N, degenerating at \( K \) points. The two branches of the linear Kagome and Dirac bands cross each other at points D1 and D2 along \( \Gamma-M \) and \( \Gamma-K \) paths, respectively. To examine whether there are extra crossing points in the whole BZ, we calculated the band structures throughout the first BZ, as shown in Fig. 2(b). Clearly, the crossing points remain along \( \Gamma-M \) and \( \Gamma-K \) paths. On the other hand, bands from the \( p_x \) and \( p_y \) orbitals are located much deeper compared with the PBE results. Because the two crossing points are not high-symmetry points, we further calculated the bands around D1 and D2 using the HSE functional with a denser \( k \)-point mesh. As shown in Fig. S2(c–f), the gap size around D1 and D2 remains essentially zero within the accuracy of numerical calculations, confirming the degeneracy of both D1 and D2.

To compare the states near the two crossing bands, we calculated the charge densities of states at points D3 and D4 close to D1 as shown in Fig. 3(a), which arise from the Kagome and Dirac bands, respectively. The D3 state mainly contributed by the \( p_z \) orbitals of C exhibits a Kagome distribution, as indicated by the green hexagons. It results from the C rings connecting with each other by an edge hopping path. As shown in Fig. 1, the eight N atoms in the unit cell form a \( 2 \times 2 \) Kagome lattice.
honeycomb sub-lattice; however, no band folding feature is observed in the band structure. Nevertheless, the Dirac bands cross at a high symmetry $K$ point, indicating a bipartite “superlattice”. For the $D_4$ state, the Dirac bands can be considered as a super-honeycomb lattice consisting of two super-atoms in the unit cell labeled with the blue triangles displayed in Fig. 3(b). Each of the super-atoms has an identical atomic configuration with $C_3$ rotational symmetry without sharing common atoms. It is noted that the two sub-lattices in the Kagome–honeycomb composite lattice proposed before\(^\text{29}\) is represented by single atoms instead of a bipartite “superlattice”. Besides, the $p_z$ (on the honeycomb sub-lattice) and $s$ (on the Kagome sub-lattice) orbitals have opposite parities with respect to the in-plane mirror, which is different from the $C_9N_4$ lattice.

We note that the nodal line discovered in the previously reported 2D flat monolayer is protected by the in-plane mirror symmetry in the absence of SOC.\(^\text{20,41}\) Meanwhile, the topological index $Z_2 = 0$ indicates that the nodal line in $C_9N_4$ is topologically trivial. In order to better understand the formation of degenerate points along the high-symmetry paths $K–\Gamma$ and $\Gamma–M$, we analyze the group symmetry and parity of bands. The point group for 2D $C_9N_4$ is $D_{6h}$, so there are both one vertical mirror plane $\sigma_v$ and one $C_2$ rotation axis existing along $K–\Gamma$ and $\Gamma–M$ as shown in Fig. 4(a and b). Along $K–\Gamma$ and $\Gamma–M$, the $k$-points (except for the end points) from the two crossing bands have a $C_{3v}$ point group, and the little group has both out-of-plane mirror $\sigma_v$ and $C_2$ rotational symmetry along $K–\Gamma$ and $\Gamma–M$. As shown in Fig. 4(d), the two linear bands along $K–\Gamma$ and $\Gamma–M$ belong to two different 1D irreducible representations ($A_2$ and $B_2$). The two bands have opposite parities either with respect to $\sigma_v$ or $C_2$ operation. Therefore, these two crossing bands along $K–\Gamma$ and $\Gamma–M$ are decoupled leading to two degenerate points, which are protected by the $\sigma_v$ mirror or the $C_2$ rotational symmetry. Such degeneracy along high-symmetry paths is similar to the case of graphene,\(^\text{46}\) and the main difference is that the touching points in 2D $C_9N_4$ are not at the high-symmetry points.

In addition, two more test calculations were carried out to artificially tune the planar structure to buckled structures to validate the role of $\sigma_v$ mirror and $C_2$ rotational symmetries, as shown in Fig. S3(a and c).\(^\dagger\) The buckled structure in Fig. S3(a)\(^\dagger\) has the $D_{3d}$ point group symmetry. Along $K–\Gamma$ and $\Gamma–M$, the little group has $\sigma_v$ mirror symmetry and $C_2$ operation, respectively. And the two crossing bands have opposite parities with respect to $\sigma_v$ and $C_2$ along $K–\Gamma$ and $\Gamma–M$, respectively, as shown in Fig. S3(b).\(^\dagger\) The buckled structure in Fig. S3(c)\(^\dagger\) has the $C_{3v}$ point group symmetry. The corresponding band structure in Fig. S3(d)\(^\dagger\) clearly shows that there is only one degenerate point along $\Gamma–M$ protected by $\sigma_v$ mirror. However, the touching point is gapped along the $K–\Gamma$ direction because both the $\sigma_v$ mirror and the $C_2$ rotational symmetry vanish. The gap size (buckling height) is smaller (larger) in comparison with other 2D nodal-line materials containing metal atoms.\(^\text{21}\) Interestingly, it is noticed that the degeneracy of Dirac bands at the $K$ point increased due to the breaking of sub-lattice symmetry in the “bipartite” lattice, resulting in a sizeable gap $E_D$ as shown in Fig. S3(d),\(^\dagger\) which is consistent with the chemical potential.

![Fig. 4](image)

**Fig. 4** (a) The 2D crystal structure and (b) BZ of $C_9N_4$, respectively, where the red and blue dashed lines indicate the out-of-plane $\sigma_v$ mirror plane and $C_2$ rotation axis along $K–\Gamma–M$. (c) Gap size obtained along $\Gamma–X(i = 1, 2, 3, 4, 5)$. (d) Band structure of $C_9N_4$ along $K–\Gamma–M$, where the “+” and “−” symbols indicate the parities of crossing bands with respect to the $\sigma_v$ mirror plane or $C_2$ rotation axis in the same color. (e) Band structure of $C_9N_4$ along $\Gamma–X3$, where the “−” symbol indicates the parities of crossing bands with respect to the horizontal mirror plane. The green and red circles are proportional to the contributions from the $p_z$ orbitals of N and C, respectively.
difference between the two super-atoms occupying the two sublattices, respectively.

In addition to the degeneracy along high-symmetry paths, we also examined other touching points in the nodal ring (Fig. 2(c)). The band structure shown in Fig. 4(e) is calculated along the $\Gamma$–X3 path. The point X3 is in the middle of the M–K' path as shown in Fig. 4(b). The magnified band in Fig. 4(e) shows that there is a tiny gap (3.5 meV) opened at the nodal point along the $\Gamma$–X3 path, which can be understood by symmetry analysis. The $k$-points along $\Gamma$–X3 (except for the end points) have the $C_s$ point group symmetry. The little group has one horizontal mirror plane $\sigma_h$. Because the crossing bands are all contributed by $p_z$ orbitals, they become coupled with the same odd mirror parities with respect to $\sigma_h$. This gives rise to the as-formed tiny gap. Tiny gaps are also found along $2D$ C$_9$N$_4$ can still be considered as a NLSM.

The above calculations are performed without SOC, so it is natural to ask whether this nodal line in C$_9$N$_4$ is immune to SOC. It is found that the SOC strength (even artificially tuning larger) induced gap size is very tiny as shown in Fig. S4.† Hence, the nodal line in C$_9$N$_4$ is robust against the intrinsic SOC effect in C and N. Therefore, all three factors, the protection of multiple crystal symmetries, negligible SOC and negligible gap size induced by $p_z$-orbital coupling, contribute together to provide the 2D C$_9$N$_4$ with ideal semimetallic properties.

To better understand the effect of the chemical potential difference on the formation of the nodal loop in C$_9$N$_4$, a TB analysis is shown below. Due to the fact that only the $p_z$ orbitals of both C and N atoms contribute to the states close to the Fermi level, we can construct a TB model with only the $p_z$ orbitals,

$$\hat{H} = \sum_i \epsilon_i a_i^+ a_i + \sum_{i \neq j} t_{ij} a_i^+ a_j$$

where $a^+$ and $a$ are electron creation and annihilation operators, $i$ and $j$ are the sites of atoms in the unit cell, $t_{ij}$ is the hopping energy between $p_z$ orbitals at sites $i$ and $j$, and $\epsilon$ is the on-site energy. We only consider the nearest-neighbor hopping without SOC. As mentioned above, there are two different C and N atoms according to the symmetry and chemical environment. Consequently, four interatomic hopping terms are introduced into the TB model, as shown in Fig. 5(a). The red and gray balls represent N and C atoms, respectively, whose size denotes the charge value based on the grid-dependent Bader charge analysis.† The calculated results indicate that each N(1) (N(2)) atom gains 1.1 (1.2) electrons, while each C(1) (C(2)) atom loses 0.6 (0.3) electrons. Apparently, the charge transfer occurs from C to N due to their electronegativity difference. The TB band structure is calculated to fit the DFT results as shown in Fig. 5(b), where only five bands comprising three Kagome and two Dirac bands are plotted. Table 1 lists the corresponding fitting parameters of hopping and on-site energies. The bond lengths between different atoms, for which the hopping integrals are indicated from $t_1$ to $t_4$ (Fig. 5(a)), are 1.39, 1.46, 1.42, and 1.34 Å, respectively. The hopping integrals are about the same since the bond lengths are close to each other. However, it is obvious that the on-site energy difference between C and N is rather large. Interestingly, we found that the superposition of

Fig. 5  (a) Bader charge analysis of the C$_9$N$_4$ unit cell. The red and gray balls represent N and C atoms. The sizes of balls indicate the charge values. (b) Band structure of C$_9$N$_4$ simulated by TB analysis (blue solid line) compared with the DFT results (red dashed line). (c) Band structure of C$_9$N$_4$ simulated by TB analysis with $\epsilon_{N(2)} = -5$ eV and (d) $\epsilon_{N(2)} = -1$ eV, while other parameters are fixed.
the three Kagome bands and the two Dirac bands is mainly attributed to the large on-site energy of N(2) atoms. As illustrated in Fig. 5(c and d), the Dirac point shifted upwards above the flat Kagome band when the on-site energy of N(2) atoms is decreased, while the other TB parameters are fixed. Moreover, the width and dispersion of the three Kagome bands are barely affected, and only the Dirac bands are influenced by the change of the on-site energy of two super-atoms in the unit cell. In particular, when the on-site energy of N(2) atoms changes to −1 eV, the linear band crossing disappears as shown in Fig. 5(d). Therefore, the linear crossing is mainly determined by the larger on-site energy of N(2) atoms. Also seen from Fig. 5(b), the band structures obtained by TB and DFT calculations agree well except for the flat band. From the DFT results (Fig. 2(a) and 5(b)), the flat band is pulled down to the K point and becomes slightly dispersive, which is different from the perfect flat band in the TB results. This is because in the TB calculations, only nearest-neighbor hopping terms are considered. In fact, such a localized flat band in the Kagome lattice is known to become dispersive when introducing second nearest neighbor hopping interactions. Furthermore, in the DFT calculations, the linear-crossing bands are contributed by the \( p_z \) orbitals of both honeycomb and Kagome lattices. Hence, there are hybridizations between the Dirac and Kagome bands. Nevertheless the main feature of linear crossing pertaining to the two sets of dispersive bands is captured by the TB model.

To further understand the formation of the intrinsic nodal ring in our 2D covalent system, the properties of molecular orbitals (MOs) for a single super atom (C_9N_4) and whole unit cell (C_9N_4) are calculated with the Gaussian package. As seen from Fig. S5, the unoccupied MOs from Dirac bands and partially filled MOs from Kagome bands determine the electron filling from the DFT results, i.e., nearly unoccupied Dirac bands and 2/3 filled Kagome bands. Moreover, the interaction between the wider dispersive Kagome bands and the narrower Dirac bands leads to a crossover giving rise to a degenerate nodal loop. More details are shown in the ESI.

The relative energy stability among different stacking configurations is investigated. We started with the AA-stacking, which has an optimized interlayer length of 3.45 Å, typical of a van der Waals (vdW) interaction. As shown in the inset of Fig. S6(a), the top layer is then shifted along one of the shorter diagonal directions until reaching half of the lattice constant of AA-C_9N_4 to assume an AB-stacking. The AB-stacking is found to be energetically more preferable by 38 meV per atom. This implies that the AA-stacking is metastable but may still be energetically accessible as seen in other 2D materials.

Due to the missing of some crystal symmetries along M–Γ–K in AB-stacked C_9N_4, some crossing points become gapped as shown in Fig. S6(b), whereas they remain gapless along M–Γ–K in AA-stacked C_9N_4, in which the crystal symmetries are preserved as in 2D C_3N_4. We note that neither AA nor AB-stacked C_9N_4 is a semiconductor, which is different from the AA and AB-stacked g-C_3N_4.

The band structure of AA-stacked C_9N_4 is shown in Fig. S7. Interestingly, the two linear crossing points shift downwards from the \( k_z = 0 \) to the \( k_z = 0.5 \) plane. Therefore, the crossing points will form a new structure of the nodal line, i.e. a “nodal cylinder”, along the \( k_z \) direction in the whole BZ, which is different from other simple nodal-line semimetals,\(^{23,24,27,37,64}\) nodal-chain metals,\(^{32}\) nodal-net semimetals,\(^{26,30}\) nodal-link semimetals,\(^{19,65,66}\) and nodal-knot semimetals.\(^{67}\) As shown in Fig. S8, there are two nodal lines along the \( k_z = 0 \) plane, which further confirm the existence of the nodal cylinder. Therefore, the preserved crystal symmetry and weak vDW interactions ensure the existence of the nodal cylinder in AA-stacked C_9N_4.

Finally, to evaluate the stability of such a new type of C–N network, we compare the energy of monolayer C_9N_4 with those of pristine graphene and other experimentally realized 2D C–N materials (C_3N, C_2N, and g-C_3N_4). The lower energy shown in Fig. S9 for monolayer C_9N_4 partially confirms its higher stability. This result is in agreement with those of the previously studied 2D C_3N, C_2N, and g-C_3N_4.\(^{6}\) However, such a comparison of total energy is insufficient because of their different stoichiometries. Therefore, we further calculate the formation energies of different C–N compounds using the chemical potentials from two known reagents, graphene and molecular nitrogen, according to the formula:

\[
\Delta E = [E_{\text{tot}(\text{C}_9\text{N}_4)} - p/2 \times E_{\text{tot}(\text{graphene})} - q/2 \times E_{\text{tot}(\text{N}_2)}]/(p + q),
\]

where \( p \) and \( q \) are the number of C and N atoms in the unit cell, respectively; \( E_{\text{tot}(\text{C}_9\text{N}_4)}, E_{\text{tot}(\text{graphene})}, \) and \( E_{\text{tot}(\text{N}_2)} \) are the total energy of 2D C–N compounds, graphene, and N_2, respectively. The formation energies in Fig. S9 show similar trends to the total energies, suggesting the energetic stability of 2D C_9N_4.

Moreover, the phonon and AIMD computations are performed. There is no soft mode observed in the phonon spectrum (Fig. S10†), indicating its dynamic stability. Further calculations are conducted to study the thermal stability at room temperature (300 K) by performing AIMD simulations. The results (Fig. S10(b and c)†) indicate that there is no structural decomposition observed at 300 K after 10 ps. We also calculated the linear elastic properties from the strain-energy relationship using DFT calculations to investigate the mechanical stability. Due to symmetry, the 2D hexagonal crystal has two independent elastic constants \( C_{11} \) and \( C_{12}.\(^{44}\) The computed values of \( C_{11} \) and \( C_{12} \) are 208 and 49 N m\(^{-1}\), respectively, which fulfill the Born stability criteria\(^{49}\) \((C_{11} > 0, C_{12} > 0)\). Other elastic quantities, such as Poisson’s ratio and Young’s modulus, can be evaluated using the following equations,\(^{70,71}\)

\[
y = \frac{C_{12}}{C_{11}} E = \frac{C_{12}^2 - C_{13}^2}{C_{11}}
\]

The obtained in-plane Poisson’s ratio (Young’s modulus) is 0.24 (196 N m\(^{-1}\)), which is larger (smaller) than those of graphene and C_3N.\(^{73}\) Therefore, our calculations of phonon...
dispersion, AIMD simulations, and mechanical elastic properties consistently suggest that the C\textsubscript{9}N\textsubscript{4} network has rather high thermodynamic and mechanical stabilities. All the evidence suggests that the synthesis of 2D C\textsubscript{9}N\textsubscript{4} under ambient conditions is possible. We propose that C\textsubscript{9}N\textsubscript{4} could be possibly synthesized utilizing similar synthesis processes applied for g-C\textsubscript{3}N\textsubscript{4,7 2} C\textsubscript{2}N,\textsuperscript{10,72} C\textsubscript{2}N\textsubscript{2}\textsuperscript{23} and COFs.42

Conclusions

In summary, a new 2D NLSM is predicted using first-principles calculations, which enriches the realm of both COFs and C-N compounds. This novel monolayer with a C\textsubscript{9}N\textsubscript{4} stoichiometry possesses a nodal line centered at the \textGamma point in the k-space forming a closed loop. This new feature is rarely found in other C-N compounds and COFs. The touching points along K-\textGamma-M are protected by the out-of-plane mirror and C\textsubscript{2} rotational symmetry. The nodal line is contributed by the p\textsubscript{x} orbitals of C and N, as further confirmed by a single orbital TB analysis. From the analysis on charge transfer and TB results, it is found that the C and N have a large on-site energy difference; particularly, the much larger on-site energy of N(2) atoms plays a significant role in forming such linear crossing at the Fermi level. The single nodal loop in C\textsubscript{9}N\textsubscript{4} is intrinsically immune to the SOC effect due to the super light constituent elements, C and N, which shows ideal semimetallic properties compared with heavy metal-involved NLSMs. Interestingly, a nodal cylinder forms in the AA-stacked C\textsubscript{9}N\textsubscript{4} along the k\textsubscript{z} direction in the whole BZ. The thermodynamic and mechanical stabilities of 2D C\textsubscript{9}N\textsubscript{4} are guaranteed by our calculations. Our results not only shed light on searching for a new kind of 2D C-N material, but also suggest a novel member among COFs with unusual semimetallic properties.

Conflicts of interest

There are no conflicts of interest to declare.

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